

Metal Oxide Nano-Array Catalysts for Low Temperature Diesel Oxidation

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June 8, 2017
@ DOE-VTO Annual Merit Review Meeting
Washington D.C.

Project ID #: ACS095

Project Overview

Overall objective:

Develop a unique class of cost-effective and high performance metal oxide based nano-array catalysts for low temperature CO and HC oxidation, 90% conversion at 150 °C or lower

Timeline

- Project start date: 10/01/2014
- Project end date: 12/31/2017
- Percent complete: ~80%

Budget

- Total project funding
 - DOE share: \$1,450,000
 - Contractor share: \$380,139

Barriers

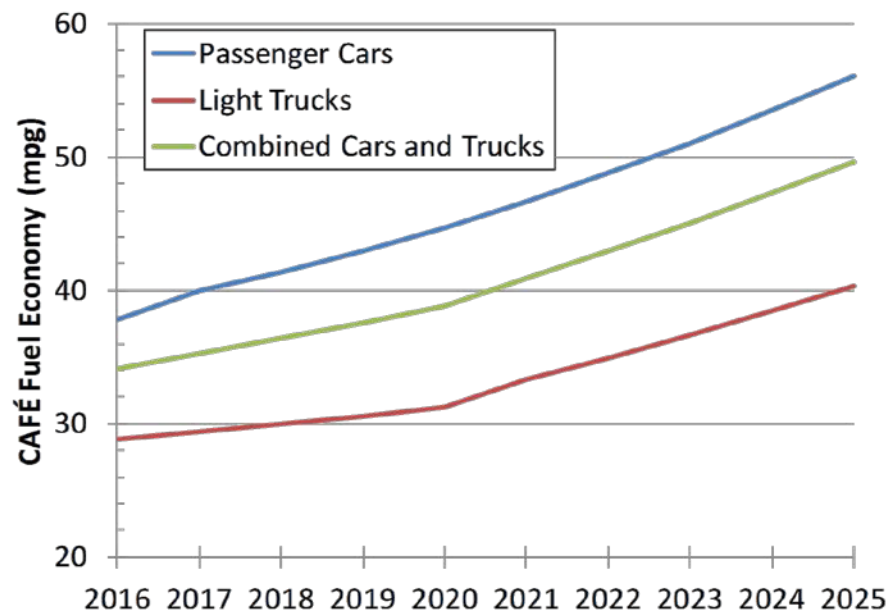
- Barriers addressed
 - From DOE Vehicle Technologies Multi-Year Program Plan
 - 2.3.1.B: Lack of cost-effective emission control
 - 2.3.1.D: Durability
 - Responsive to USDRIVE ACEC Tech Team Roadmap, Low Temperature Aftertreatment Workshop Report

Team Partners

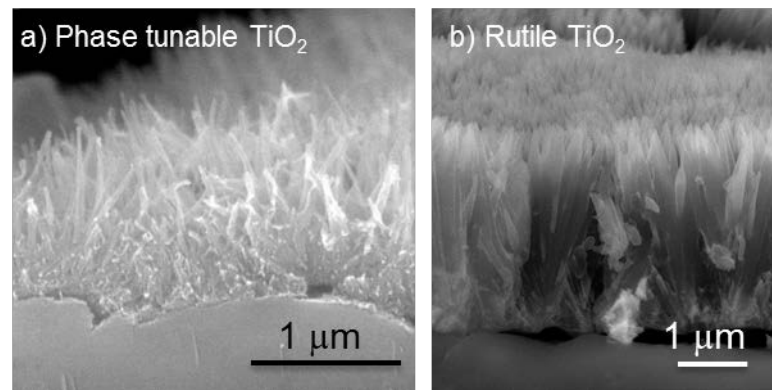
- ORNL, Umicore, and 3D Array Technology LLC

Project Relevance

- Improved fuel economy standards will require advanced combustion engines with greater fuel efficiency and consequently **low exhaust temperatures**
 - Challenges:
 - Stricter emissions standards
 - Greater HC + CO emissions
 - Low reactivity at 150°C
- **New technology needed**
- Investigate nano-array catalysts for low-cost pathway to 90% conversion of HC + CO at 150°C

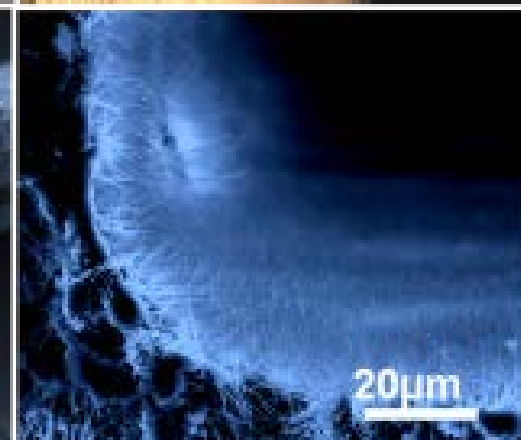
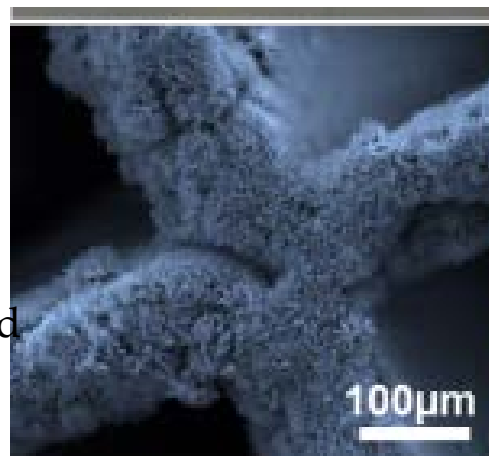
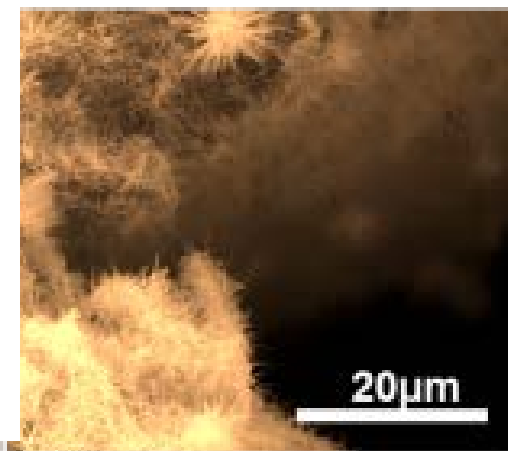
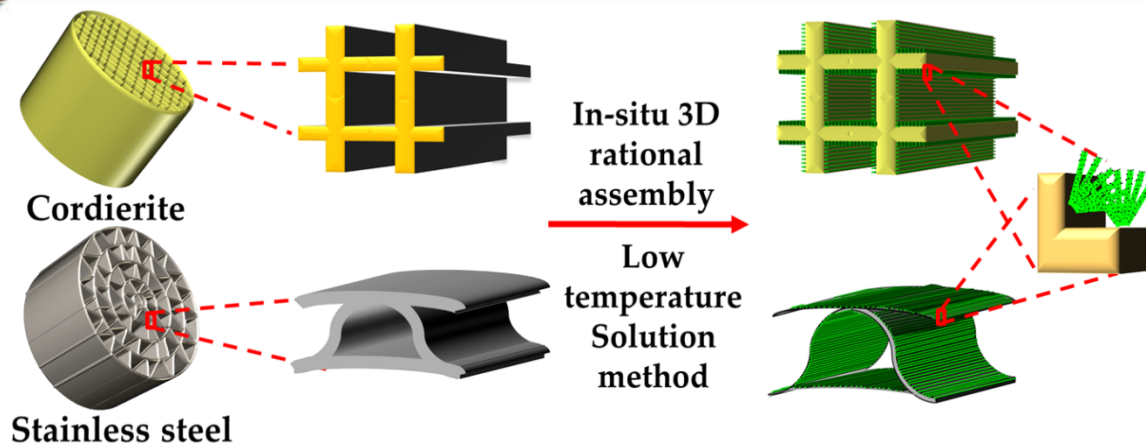


C. DiMaggio, "ACEC Low Temperature Aftertreatment Program", 6/21/12.



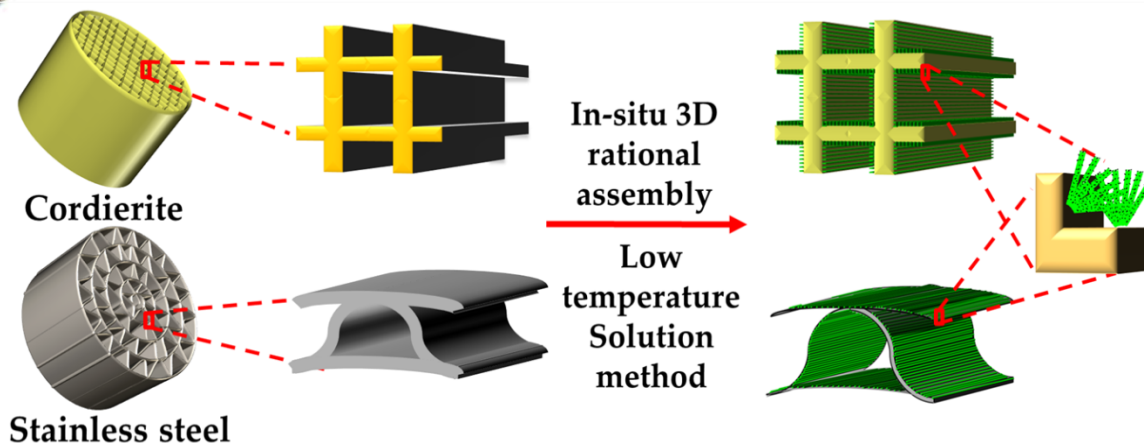
Low-temperature processed and hydrothermally stable TiO₂/Pt based nano-array catalysts: a) phase tunable TiO₂; b) mesoporous rutile TiO₂.

In-situ Growth of Nano-arrays onto Honeycomb Monoliths



- In-situ solution-based growth of (nanostructure array) nano-array on monolith
- Free of binders, robustness due to strong substrate-array adhesion
- Reduced PGM and other materials
- Improved efficiency due to size, shape, and structure
- Hydrothermally stable (e.g., TiO_2 , CeO_2 , Al_2O_3)

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- **Demonstrated on a range of scales**

Responses to AMR comments

(Project period: 4/1/2016-03/31/2017)

- **Approach** very unique is the growth of nano-array catalysts in an existing honeycomb support... In order to downselect appropriate materials to advance to the next level, **realistic test conditions and aging treatments must be employed sooner**. ...the in-situ growth of nano-array catalysts on monoliths is very interesting, although the reviewer wondered how relevant this would be commercially. ... **how possible is the scale-up of this coating method, how durable are the nano-arrays, and are they more susceptible to thermal stresses, sintering, and aging**. The reviewer pointed out **the need to have assessed S poisoning effects** by now (based on comments from last year).

Responses to AMR comments

(Project period: 4/1/2016-03/31/2017)

- **Approach** very unique materials to advance of nano-array catalysts
how possible is the stresses, sintering, and
comments from last

Responsive Actions
<i>Focus on testing under realistic simulated gases and mixture gases for species inhibition/promotion understanding</i>
<i>Focus on evaluation of mechanical/hydrothermal stability following protocol procedures</i>
<i>Aging studies being performed (including sulfur exposure); Array design over PGM doping and CeO₂ or Al₂O₃ decoration for water/S mitigation; Surface area tracked for catalysts</i>

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- **Technical Accomplishments:** minor progress had been achieved in the characterization of multiple catalyst formulations using more realistic feed conditions and aging treatments... However, **poisoning effects were not addressed and comparison to a reference, traditional PGM catalyst was not done as a benchmark.** The reviewer noted that **from a manufacturing perspective, using a growth technique to deposit an active catalyst material on a substrate may preclude the adoption of this technology. The reviewer inquired as to whether any progress has been made to reduce this challenge** ...Interesting results have been obtained,**more work needs to have been done using the ACEC protocol compositions with HTA**, which they have begun... One concern is that the Pt size increased due to HTA, which again brings up the question as to **the nano-array stability. Current testing methods do not allow for separation of kinetic and mass transport properties** ... The reviewer noted that transition metal results on Slide 12 are of limited value because they **were collected under totally unrealistic conditions.**
- **Future plans:** The reviewer noted that lots of work remains and it may be necessary **to focus on the most promising one or maybe two of the catalyst systems** to get as complete a set of analyses as possible. **Some effort should be spent on understanding the role of kinetics versus mass transport.** The reviewer stated that the project team indicated last year that more progress would be made testing under realistic conditions and aging methods

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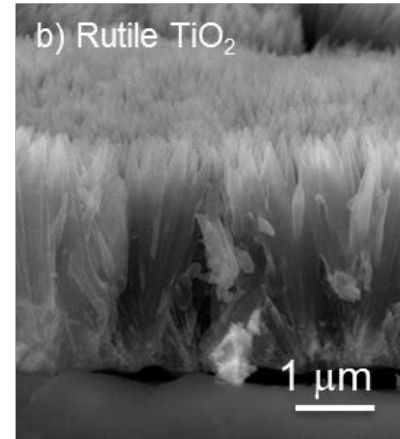
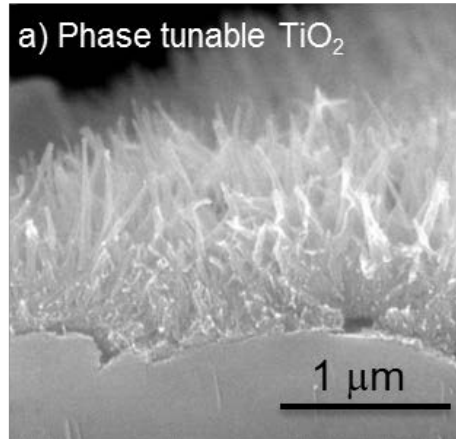
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Responsive Actions
<i>Focus on testing under realistic simulated gases and mixture gases for species inhibition/promotion understanding</i>
<i>Commercial DOC was employed as benchmark</i>
<i>Focus on scale-up synthesis; Full size substrates have been prepared for transient engine and engine test; Scalable low temperature TiO₂/Pt synthesis demonstrated</i>
<i>Kinetics of propane oxidation over Pt/TiO₂ have been studied</i>

- **Future plans:** The reviewer stated that the project team indicated last year that more progress would be made testing under realistic conditions and aging methods **two of the catalyst's role of kinetics versus mass transport.** The reviewer stated that the project team indicated last year that more progress would be made testing under realistic conditions and aging methods

TiO₂ Nano-array Synthesis Technologies

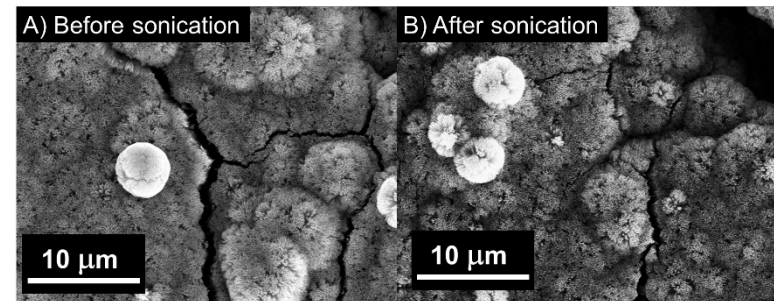
- Two synthesis/processing technologies to integrate TiO₂ nano-arrays onto ceramic monoliths: moderate (150 °C) and low (<= 90 °C) temperatures.



- Full size substrates (Φ5.66" x 3") have been prepared for heavy-duty engine test. Transient engine test performed on samples (Φ2.5" x 3")

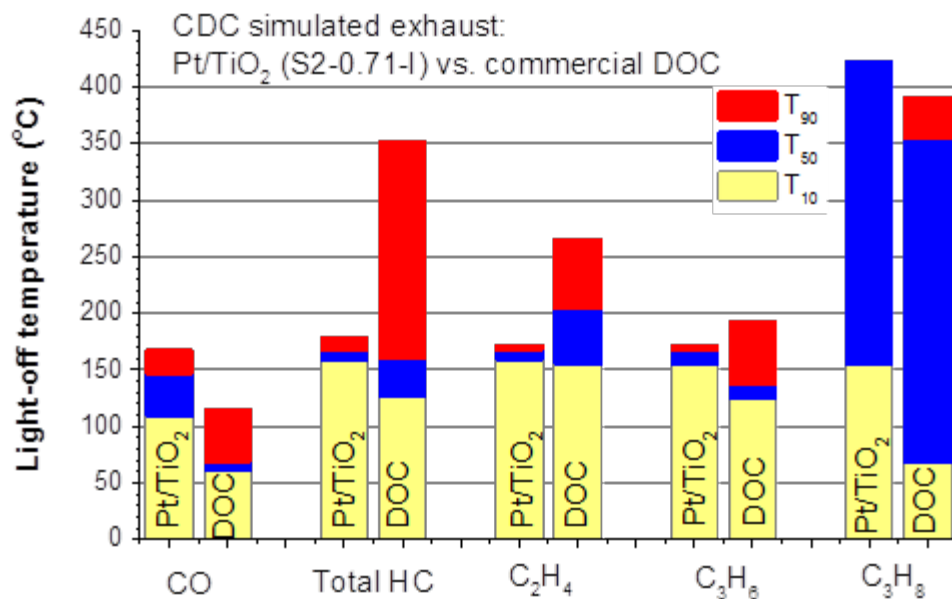


- In-situ growth results in excellent mechanical stability: negligible weight lost and morphology change after ultra-sonication test

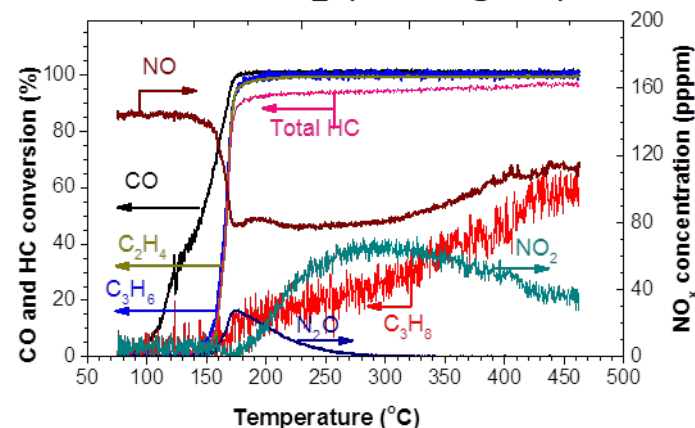


Pt/rutile TiO₂ Nano-array based Catalyst in CDC simulated exhaust

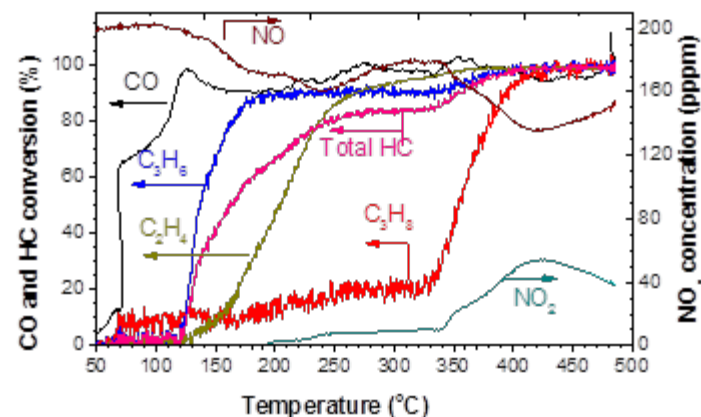
- Excellent performance of Pt/TiO₂ nano-array especially for unsaturated HC oxidation.
- T₉₀ for THC as low as 178 °C vs. 355 °C for DOC
- Propane oxidation remains challenging → more on kinetics investigation later



Pt/rutile TiO₂ (20Pt g/ft³)

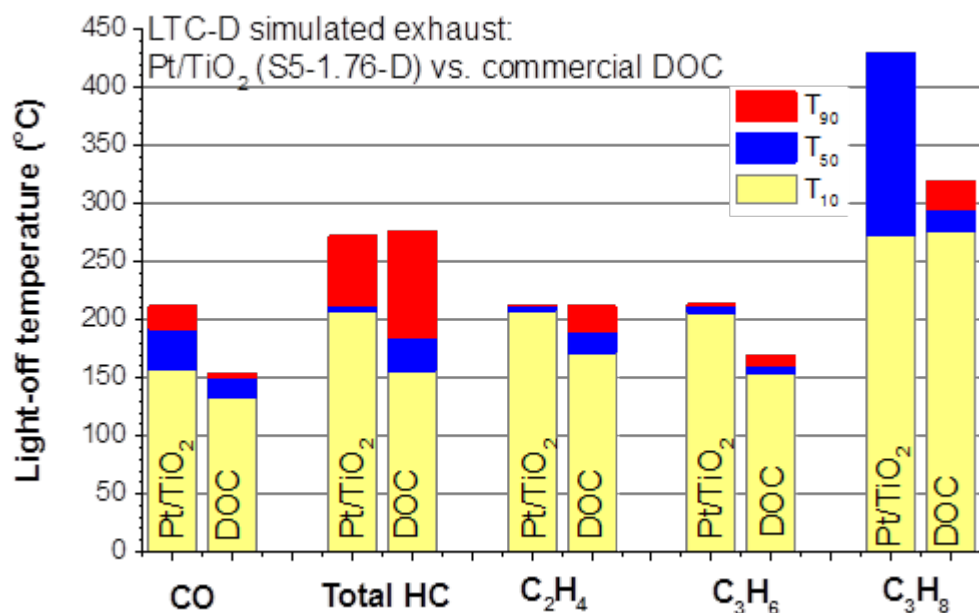


DOC bench-mark (25Pt/105Pd g/ft³)

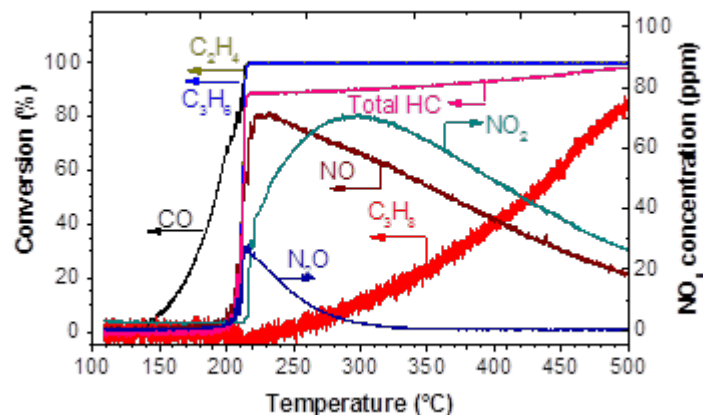


Pt/rutile TiO₂ Nano-array based Catalyst in LTC-D simulated exhaust

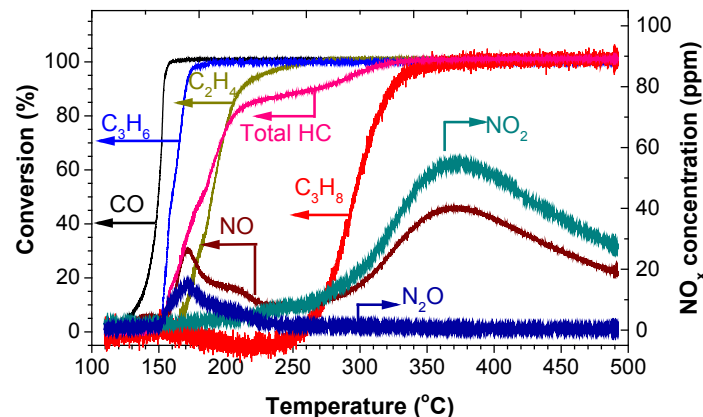
- Deactivated in LTC-D exhaust due to high concentration of CO and THC
- Comparable or even better for THC oxidation with DOC, despite 2.5 times less PGM loading
- Out-performing for NO oxidation



Pt/rutile TiO₂ (50Pt g/ft³)

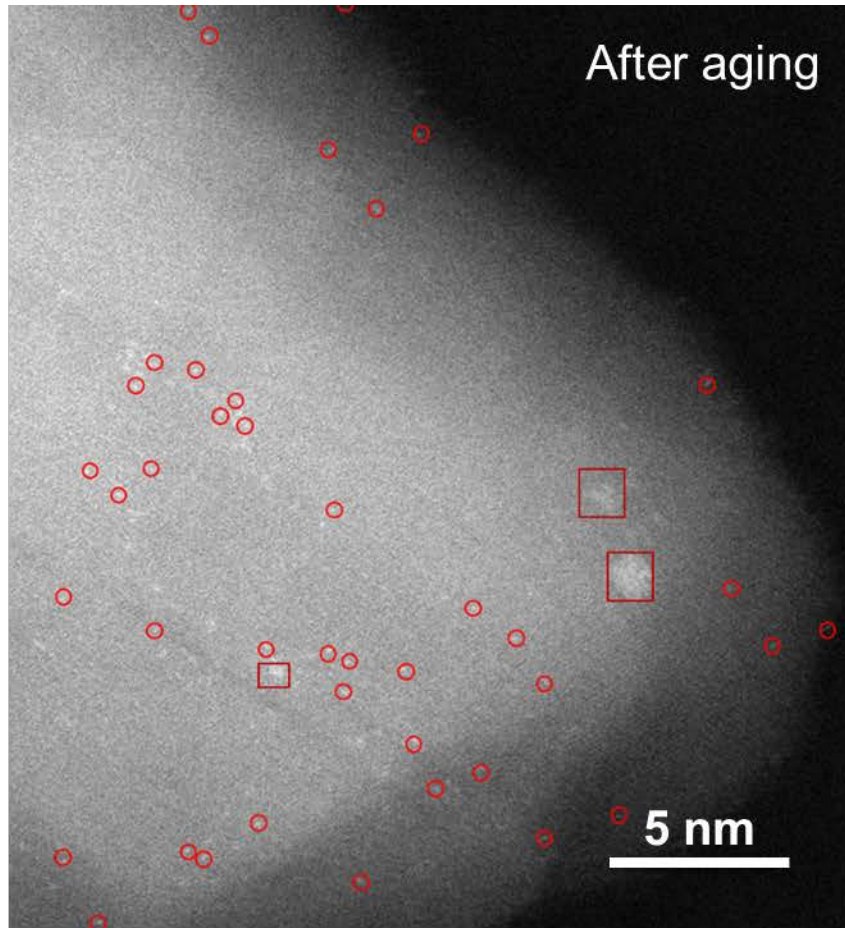
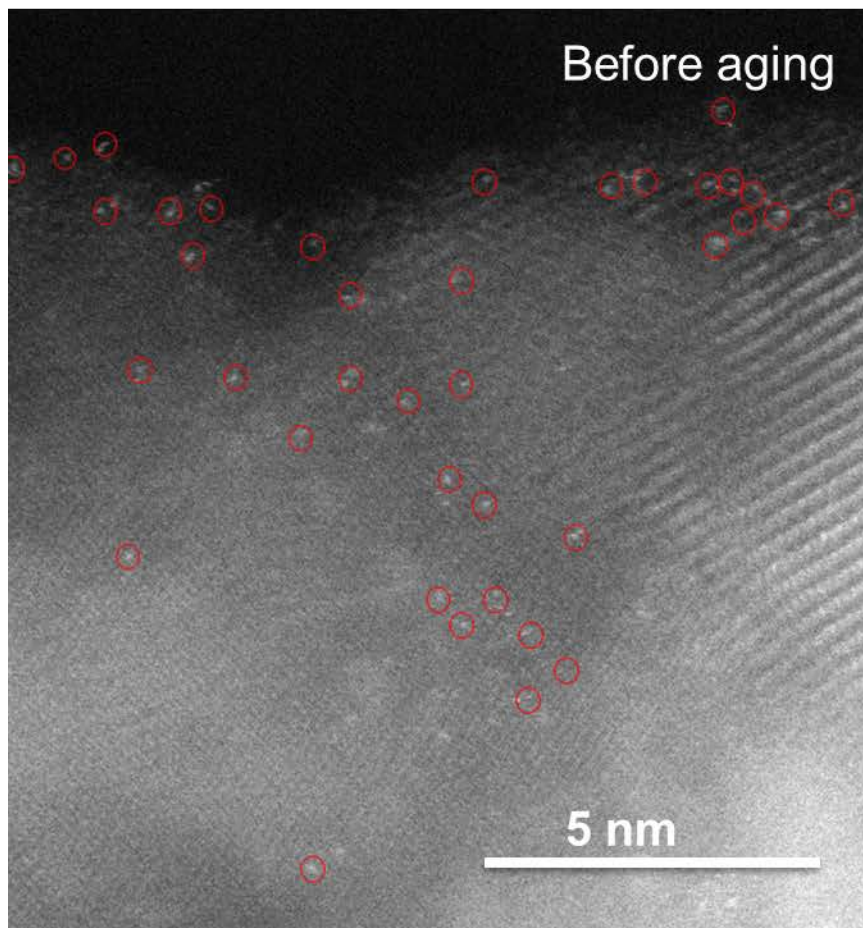


DOC bench-mark (25Pt/105Pd g/ft³)



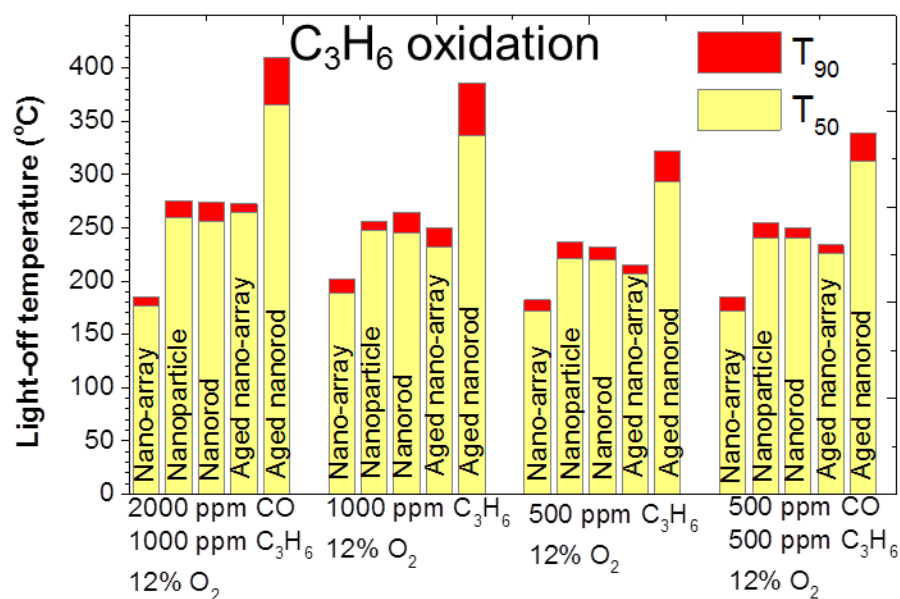
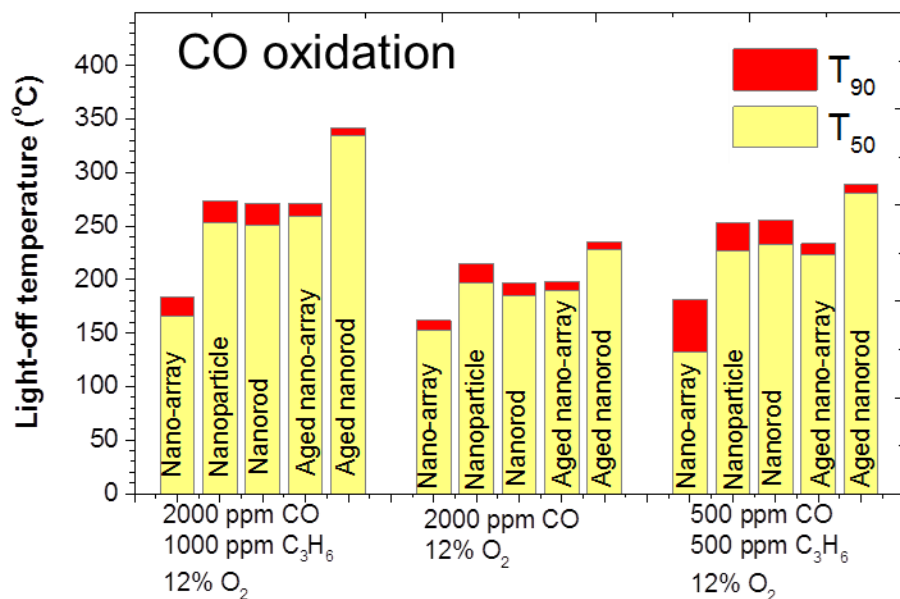
Pt/rutile TiO₂ Nano-array based Catalyst

- Unique feature: Pt was dispersed as single atoms and sub-nanometer clusters
- Pt single atoms are stable even after hydrothermal aging at 700 °C for 4 hours



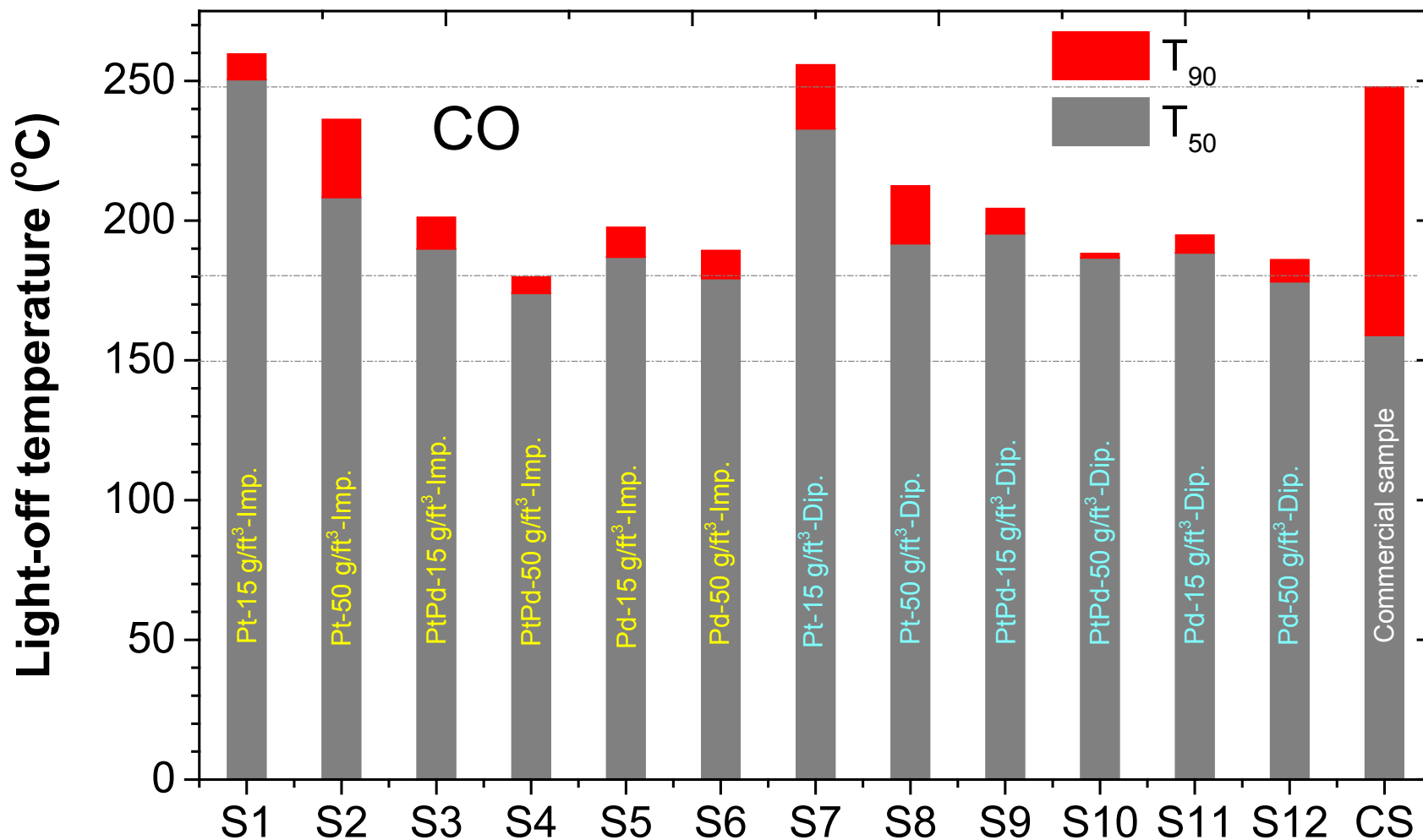
Hydrothermal stability study of Pt/rutile TiO₂ Nano-array

- CO and C₃H₆ as probe molecules
- Nano-array vs. Powder wash-coated samples: 20 g Pt/ft³, ~12.5% TiO₂
 - Pt/TiO₂ nano-array showed better performance, less inhibition effects
- Effect of hydrothermal aging (700 °C for 100 hours):
 - deactivation, but performance is still on par with fresh wash-coating samples.
 - nano-array structure remains but surface area decreased from 45 to 16 m²/g



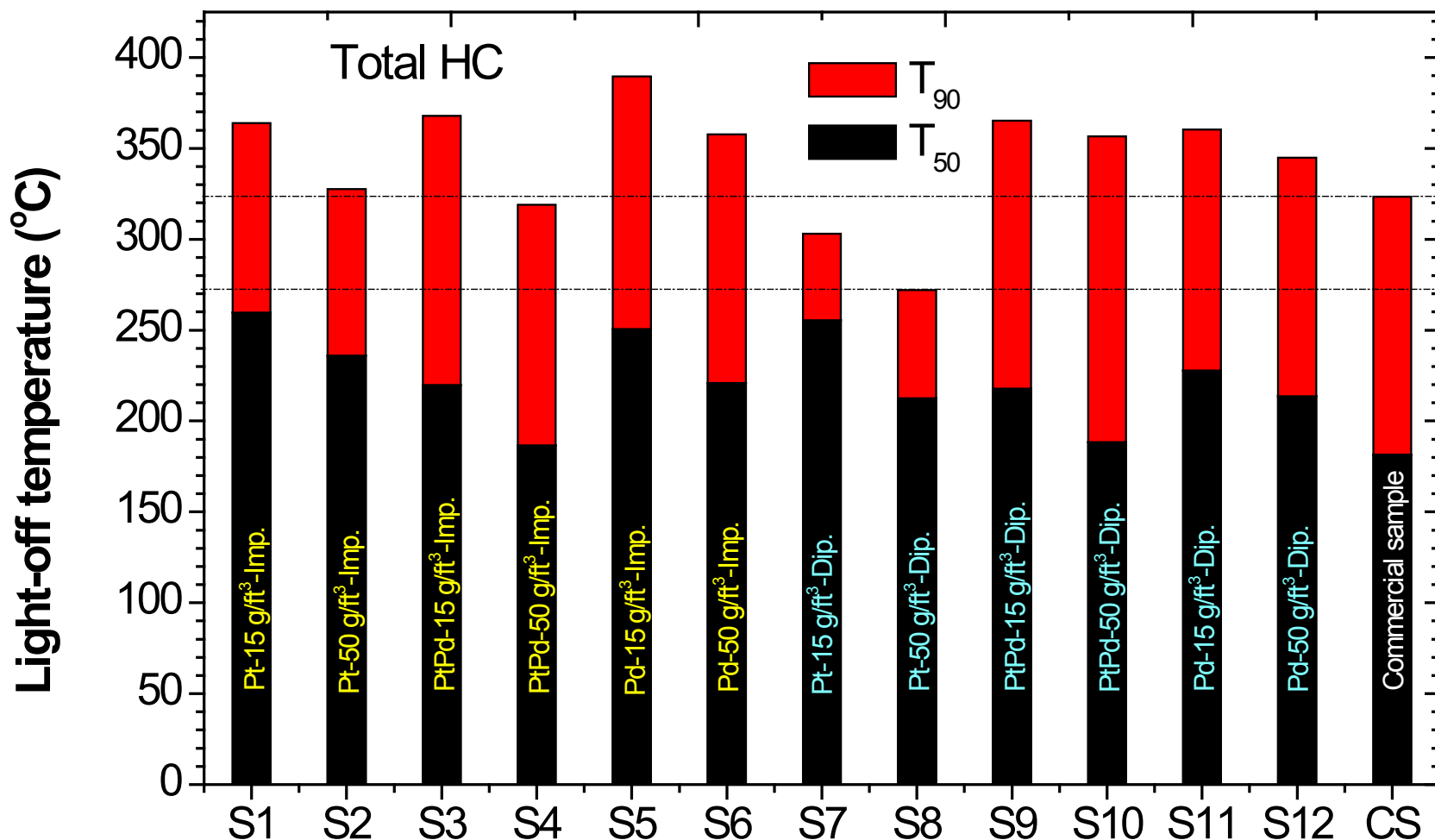
Pt-Pd/rutile TiO₂ Nano-arrays for LTC-D

- Various Pt-Pd formulations (PGM loading, 3Pt/2Pd wt. ratio) tested
- Pt-Pd alloys showed better CO and HC oxidation



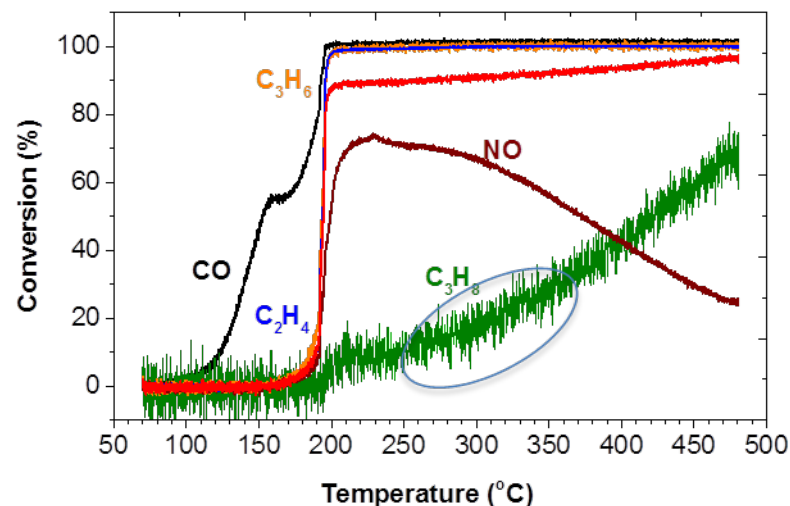
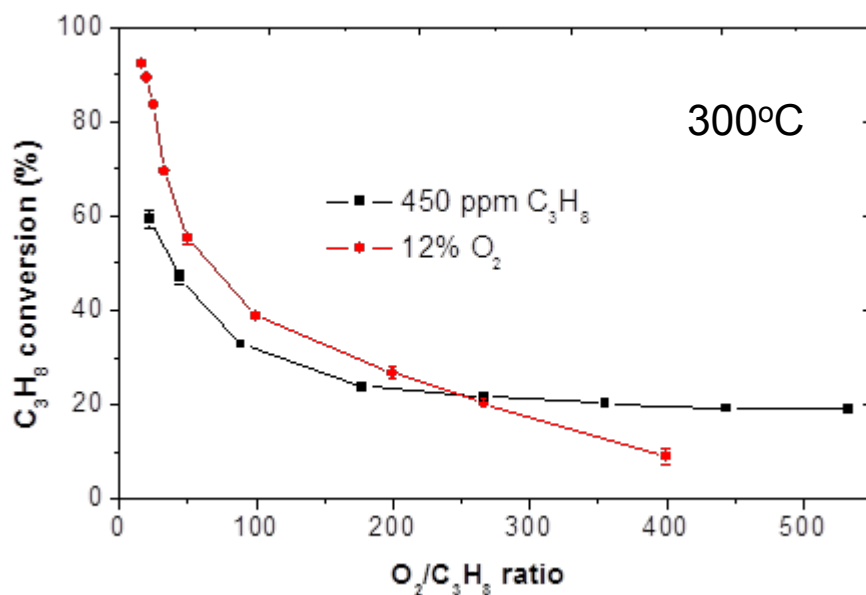
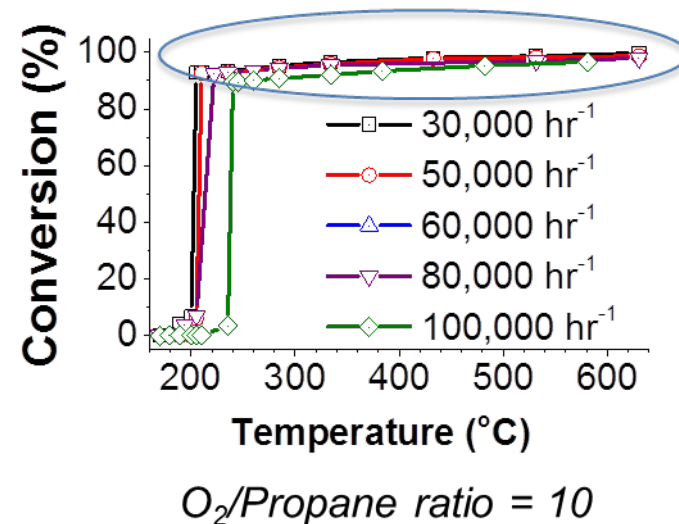
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Kinetics study of Propane oxidation over ALD Pt/rutile TiO₂ Nano-array

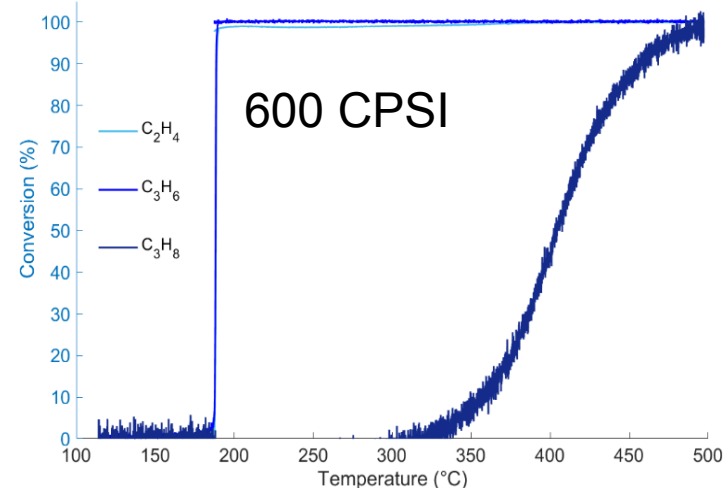
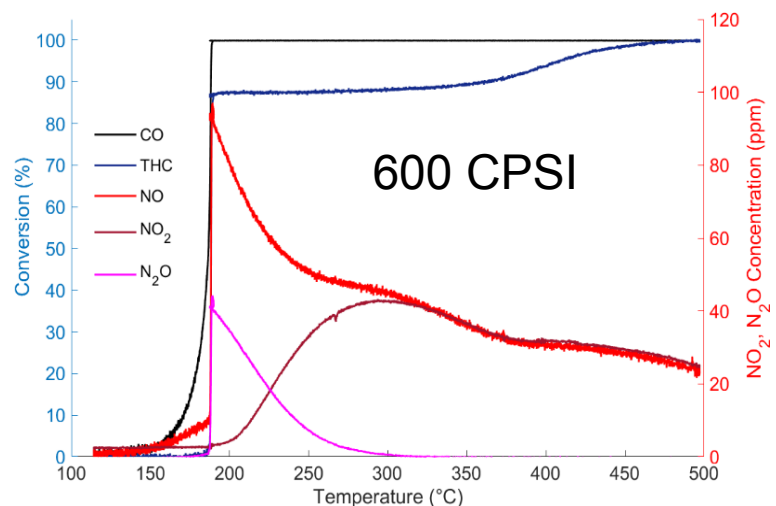
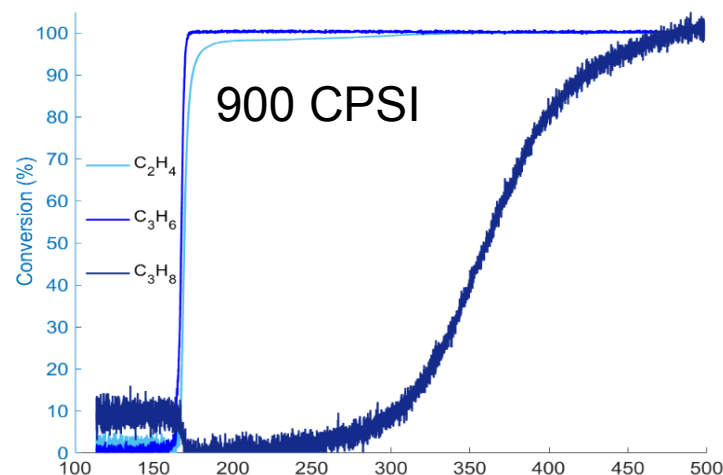
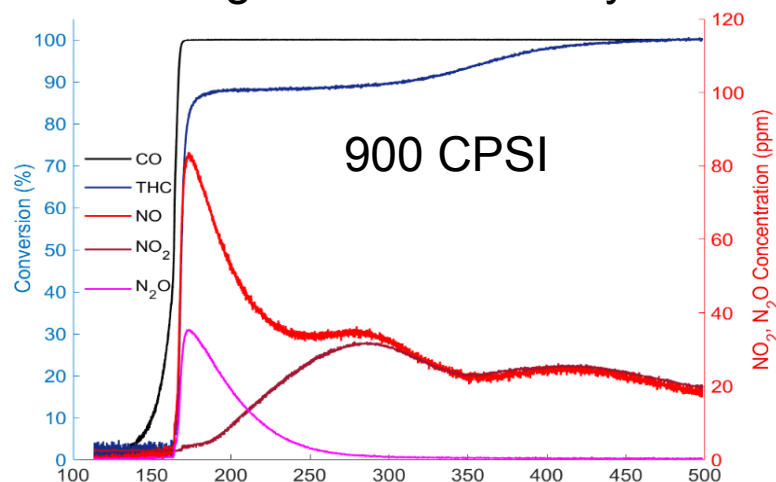
- Excellent in high propane (0.8%) but poor in the CDC condition (55.4 ppm C₃H₈)
- Almost no dependence in space velocity up until 80,000 h⁻¹ → excellent mass transport
- Plateau in high C₃H₈ conversion region
- Kinetics change due to varying O₂/propane ratio: competition for adsorption sites between C₃H₈ and O₂ & Pt oxidation states— unique for Pt catalysts → solution: Pt-Pd alloy catalysts



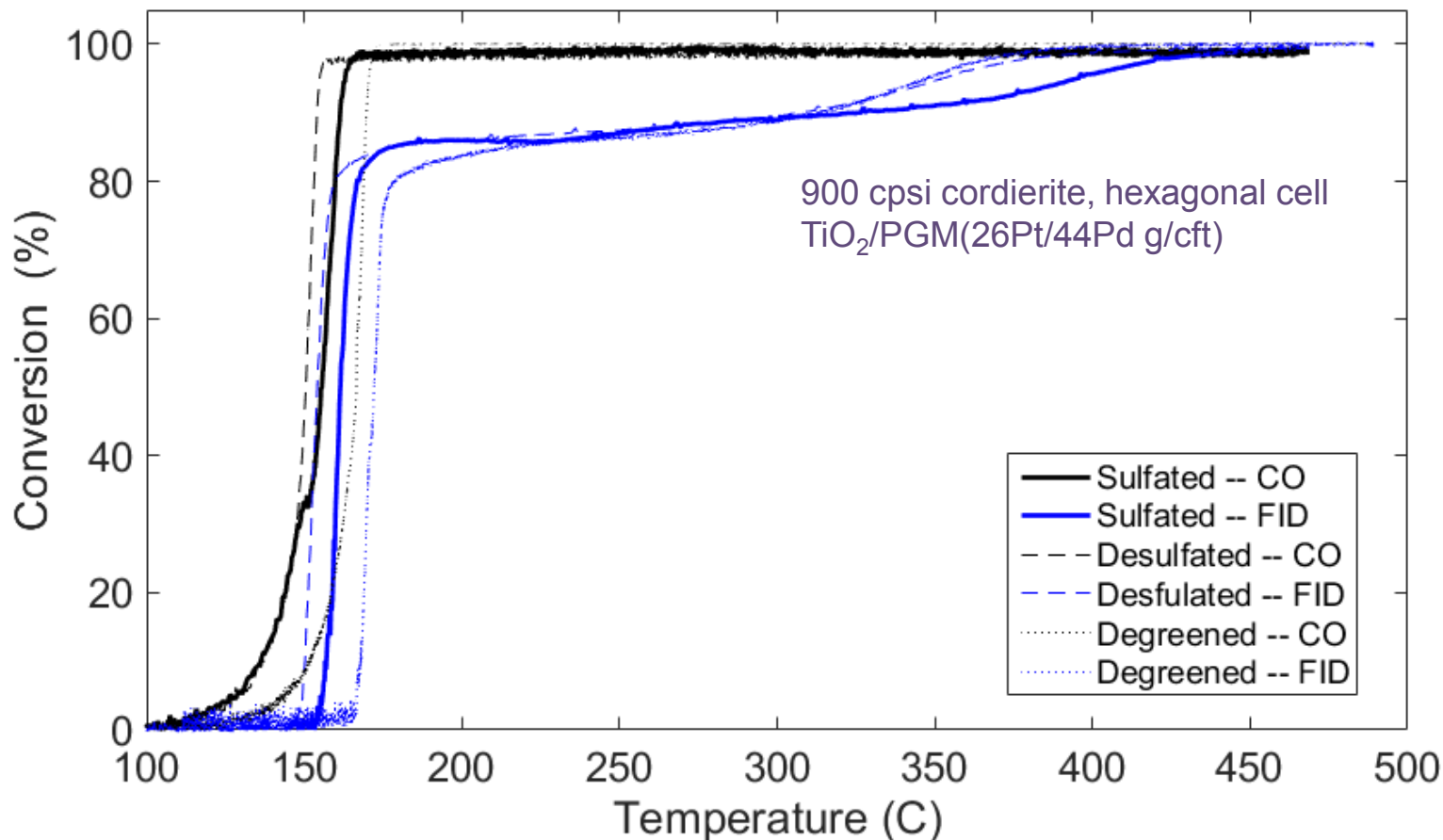
Pt-Pd/rutile TiO₂ Nano-arrays for LTC-D:

Substrate effect

- In-situ growth addressed better back-pressure issue compared to wash-coating
- Higher cell density showed better performance due to higher catalyst loading and coverage under similar synthetic conditions (30Pt/20Pd g/cft)



S-poisoning effect on Pt-Pd/rutile TiO₂ Nano-array Catalyst under LTC-D Mixture



Step 1: 300°C, 5 ppm SO₂ added to the standard LTC-D mixture to sulfate the catalyst for 5 hours.

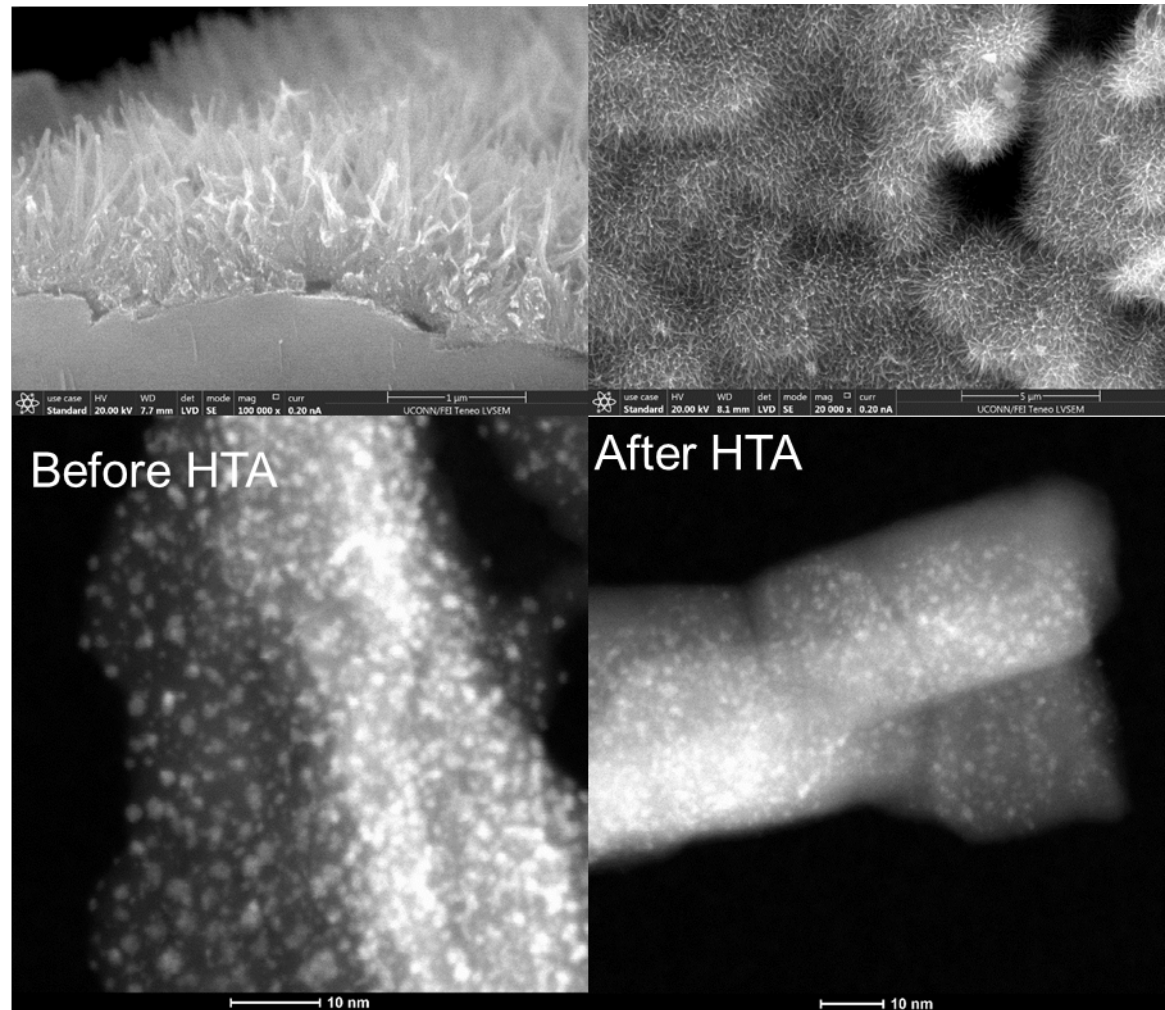
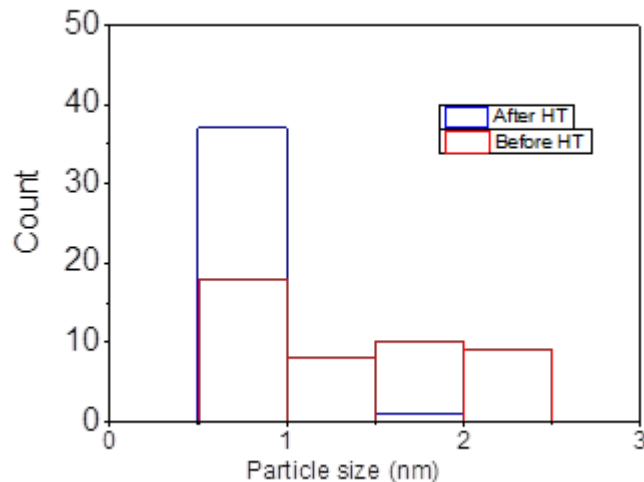
Step 2: Catalyst brought down to 100°C and ramped to 500°C for the sulfated light-off curve.

Step 3: Desulfation at 500°C by cycling 10% O₂ and 1% H₂ for 30 minutes (30 seconds per condition). 6% CO₂ and 6% H₂O in the common stream during this time.

Step 4: Catalyst down to 100°C and back up to 500°C under LTC-D conditions for desulfated light-off curve.

Pt/ TiO₂ Nano-array by Low-temperature Process

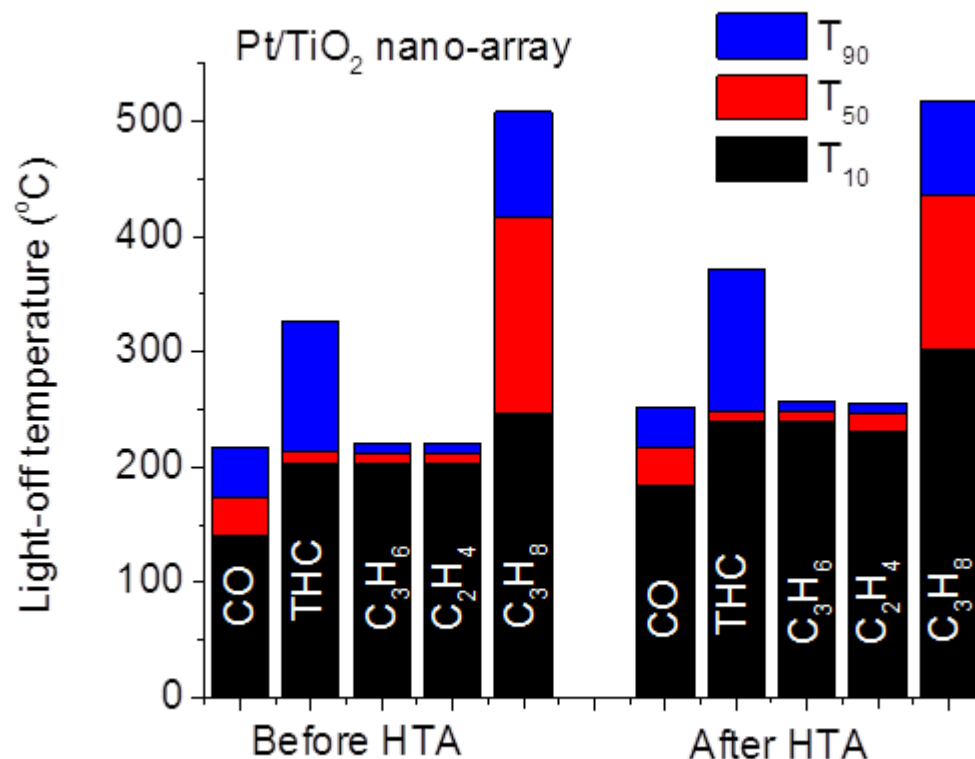
- Low processing temperature
- TiO₂ phase tunable by post-thermal treatment
- Pt particle seemed to be smaller after hydrothermal aging (700 °C, 100 h), likely with a Pt anchoring effect into TiO₂ relevant to SMSI effect



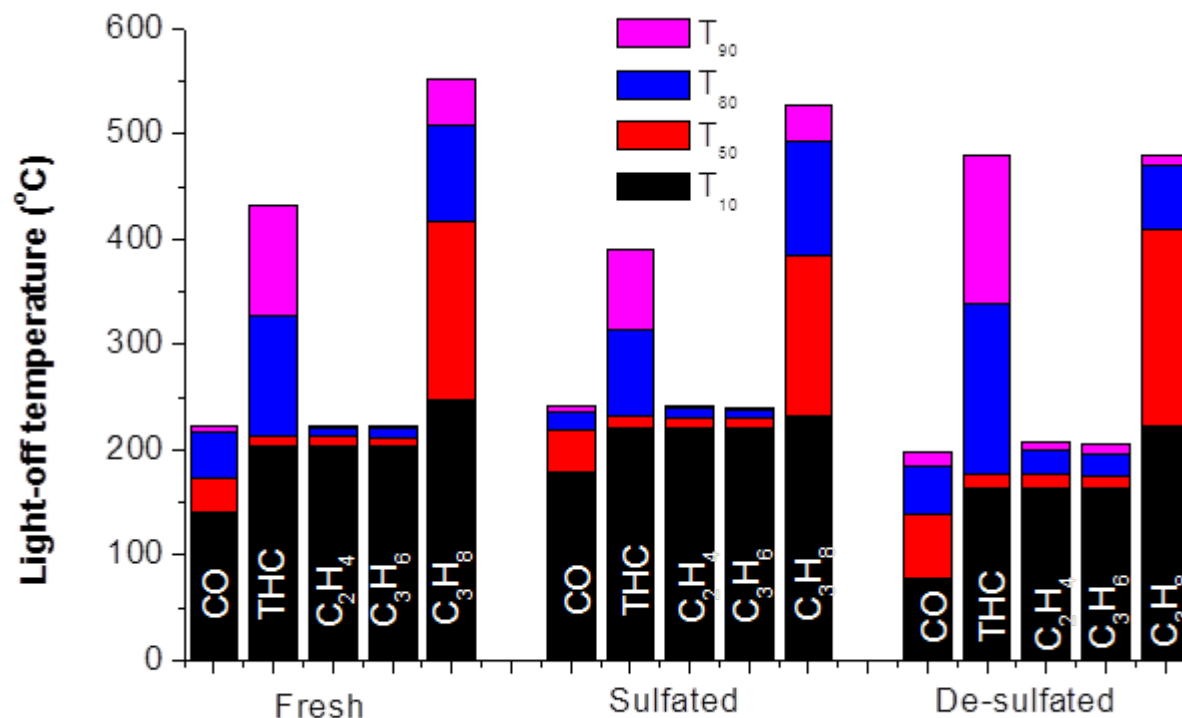
Pt/ TiO₂ Nano-array by Low-temperature Process for LTC-D: Hydrothermal stability

- Good performance with low Pt loading (50Pt g/ft³)
- Excellent HT stability: after aging at 700 °C for 100 h, T90 increases by less than 40 °C

	T10	T50	T90
CO	70	106	180
THC	163	175	330
C ₂ H ₄	163	175	196
C ₃ H ₆	163	175	193
C ₃ H ₈	220	410	470



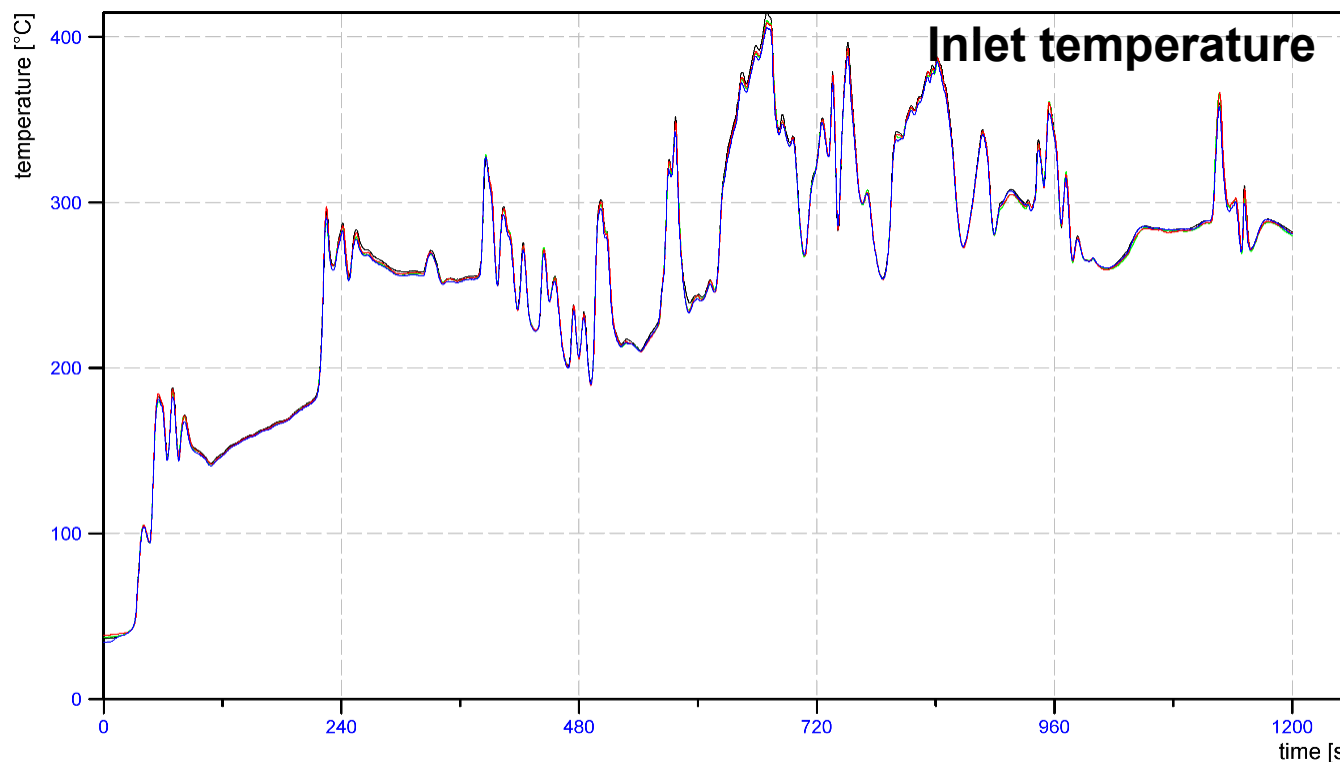
Pt/ TiO₂ Nano-array by Low-temperature Process: S poisoning Effects



- Good S-poisoning resistance (50Pt g/cft):
 - CO and unsaturated HC: T₉₀ increased less than 20 °C
 - propane oxidation is even better → lower T₉₀ for THC after sulfation
- Recovered by H₂ treatment:
 - CO and unsaturated HC: T₉₀ decreased ~ 50 °C, better than fresh sample

Pt/rutile TiO_2 Nano-arrays

High dynamic test HC emissions



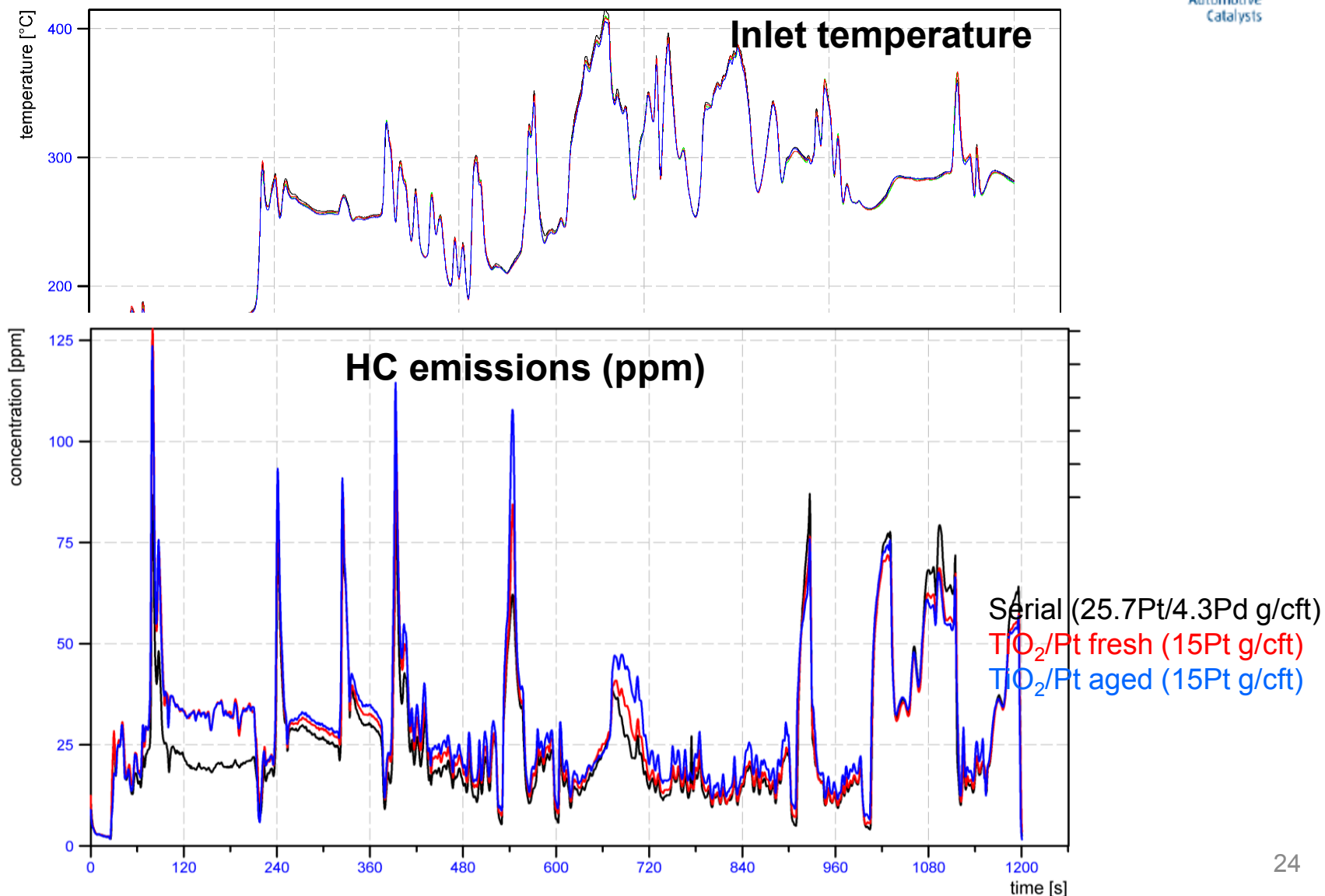
Serial (25.7Pt/4.3Pd g/cft)

TiO_2/Pt fresh (15Pt g/cft)

TiO_2/Pt aged (15Pt g/cft)

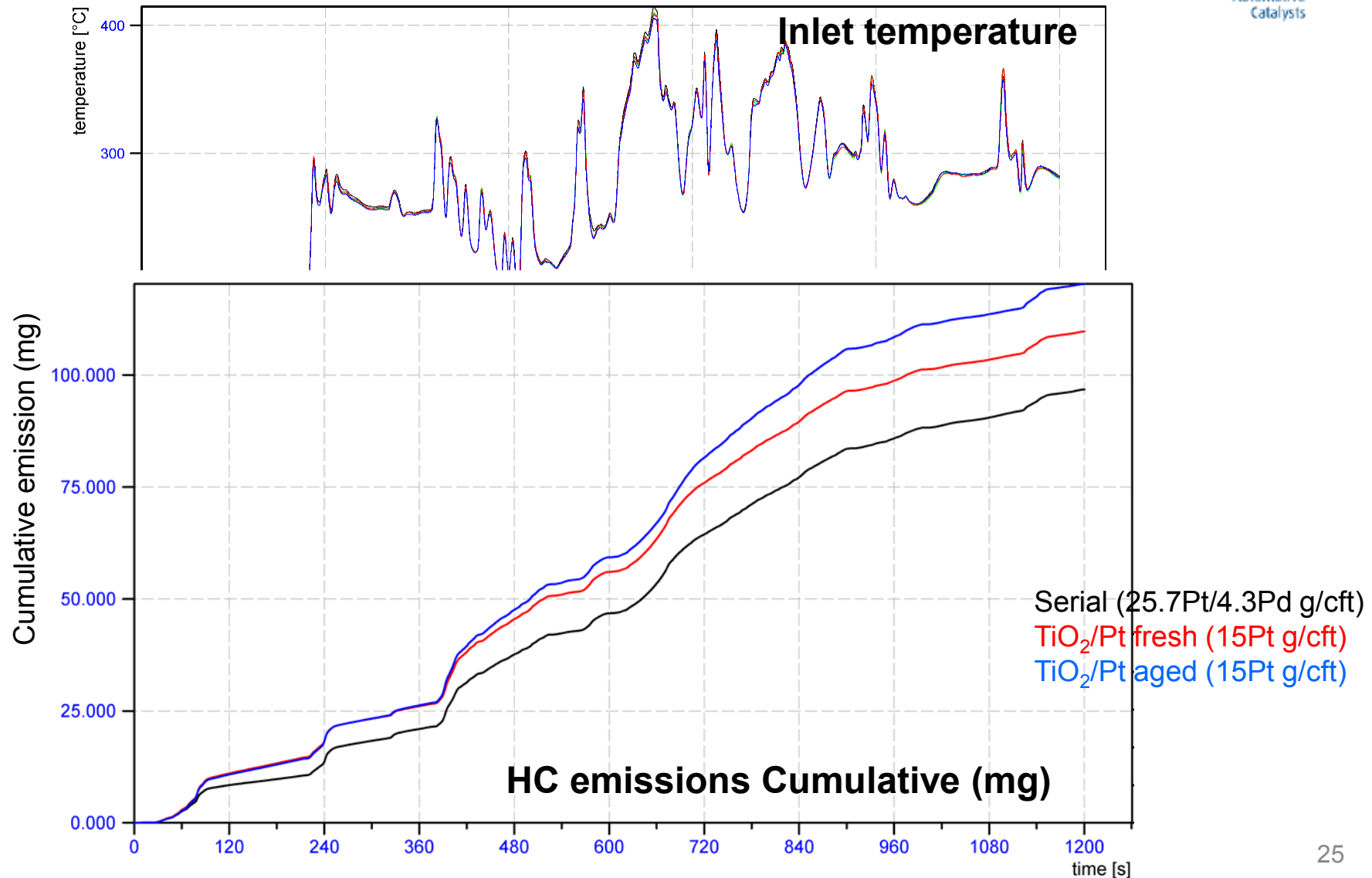
Pt/rutile TiO_2 Nano-arrays

High dynamic test HC emissions



Pt/rutile TiO₂ Nano-arrays

High dynamic test HC emissions





Engine Dynamometer Test

- 3 sets of TiO_2/Pt nano-array samples being tested at Umicore/FEV Inc., Auburn Hills, MI
- Specific Equipment:
 - i. Engines:
 - i. (MDD/HDD target) 2010 Cummins ISB6.7
 - ii. (LDD target) 2014 Chrysler Eco diesel 3.0L
 - ii. AC Dynamometer - 650HP max (Performance evaluation)
 - iii. Hydrothermal oven aging capabilities
 - v. HC injection and control

Conclusions and Future Work

1. **Approach:** in-situ growth of nano-array based catalysts onto various honeycomb substrates. Simulated exhaust gas and FTP transient cyclic conditions applied in catalytic oxidation testing.
2. **Relevance:** nano-array catalysts with low PGM and other materials usage, low temperature performance, and excellent robustness, meeting the needs of fuel economy, regulation, low temperature combustion, and environmental protection.
3. **Tech accomplishment:**
 - a) TiO_2/Pt based nano-array catalysts demonstrated excellent THC oxidation and CO oxidation performance under both CDC and LTC-D exhaust conditions. Under CDC condition, the THC 90% conversion temperatures approach 150°C .
 - b) Hydrothermally aged TiO_2/Pt nano-array structures retained both structural and catalytic performance very well.
 - c) S-poisoned TiO_2/Pt nano-array structures retained both structural and catalytic performance with even better performance after desulfation.
 - d) FTP transient cyclic test showed that nano-array catalysts displayed similar conversion efficiency of THCs to that of reference benchmark;
 - e) Large scale nano-array devices engine tests on-going.
4. **Collaboration:** ORNL, Umicore, and 3D Array Tech., and other labs (BNL, NETL) and universities (Georgia Tech, Arizona State U. and UT Dallas).
5. **Future work:**
 - a) More doping and loading studies on $\text{TiO}_2/(\text{Pt},\text{Pd})$ systems for the 90% 150°C THC conversion under CDC and LTC-D simulated exhaust conditions;
 - b) More mechanistic study on the water/S effects of TiO_2 based nano-array catalysts;
 - c) Large scale nano-array catalysts for engine dynamometer evaluation.

Acknowledgements

- Postdoc: Drs. Son Hoang, Wenxiang Tang, Andrew Binder, Eleni Kyriakidou
Graduate students: Junfei Weng, Xingxu Lu, Sibao Wang, Wenqiao Song, Sheng-Yu Chen,
Co-PIs: Drs. Steven Suib, Yanbing Guo (UConn), Dr. Todd Toops (ORNL), Dr. Tom Pauly (Umicore), Dr. Yanbing Guo (3D Array Tech.)
- Collaborators: Drs. Zili Wu, Steven Overbury, Jim Parks (ORNL), Drs. Chang-Yong Nam (BNL), Dr. Yong Ding (Georgia Tech), Dr. Jingyue Liu (Arizona State U.), Dr. Jie Zheng (UT Dallas), Drs. John Baltrus, Paul Ohodnicki (NETL)
- Industrial partners: Corning, Umicore, 3D Array Tech..
- Project officers: Ken Howden, Ralph Nine
- DOE/NETL, NSF, DOE Office of Science User Facilities program (ORNL-CNMS, BNL-CFN)